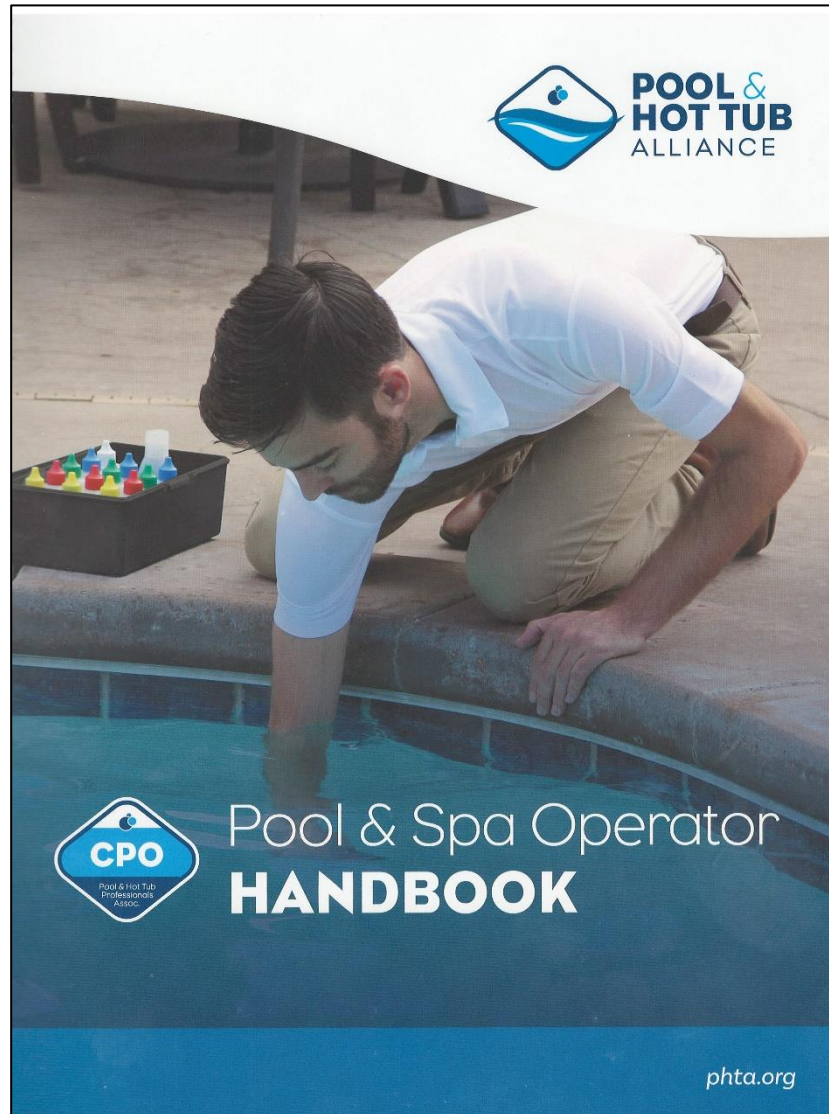


# Certified Pool & Spa Operator Certification Course

2023-r3

## Handout Packet



*Presented by:*



**cpo**.TRAINING™  
CERTIFIED POOL OPERATOR  
*A Passion for Healthy Water and Aquatic Education*

# Class Agenda

## 2-DAY CERTIFIED POOL & SPA OPERATOR (CPO) COURSE

Day 1		Chapter	Topic
A.M.	8:00 – 8:45	-	Introduction and Housekeeping
	8:45 – 9:00	1	Pool & Spa Management
	9:00 – 9:15	2	Regulations & Guidelines
	9:15 – 10:15	3	Essential Calculations
	<b>10:15 – 10:30</b>	-	<b>Break</b>
	10:30 – 11:00	4	Pool Water Contamination
	11:00 – 12:00	5	Disinfection

	<b>12:00 – 1:00</b>	-	<b>LUNCH</b>
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P.M.	1:00 – 1:45	6	Water Balance
	1:45 – 2:30	7	Pool & Spa Water Problems
	2:30 – 3:00	8	Chemical Testing
	<b>3:00 – 3:15</b>	-	<b>Break</b>
	3:15 – 3:45	9	Chemical Feed & Control
	3:45 – 4:30	10	Water Circulation
	4:30 – 5:00	11	Pool & Spa Filtration

Day 2		Chapter	Topic
A.M.	8:00 – 8:30	-	Internet– National, State and Local codes
	8:30 – 9:00	12	Heating & Air Circulation
	9:00 – 9:30	13	Spa & Therapy Operation
	9:30 – 10:15	14	Facility Safety
	<b>10:15 – 10:30</b>		<b>Break</b>
	10:30 – 10:45	15	Keeping Records
	10:45 – 11:00	16	Maintenance Systems
	11:00 – 11:15	17	Trouble Shooting
	11:15 – 11:30	18	Renovation & Design, ADA Revision of 2010

	<b>11:30– 12:30</b>	-	<b>LUNCH</b>
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P.M.	12:30– 1:30		Review, Questions, Exam Instructions
	1:30 – 4:00		Exam

## BASIC MATH SKILLS TEST



### Calculator Functions

+	= Add
-	= Subtract or Takeaway
±	= Divide
x	= Multiply
=	Represents equals

This basic math skills test is design to help you test your basic math knowledge that you will be required to perform as a CPO® graduate. This is also your chance to make sure that you are familiar with using a calculator. Complete this test before you move on to the next section. The answers to these basic math questions are found on the next page but try not to look until you have completed all the questions. Your CPO® instructor may ask to see your answers to these questions at the beginning of your CPO® course. Simply circle your answer choice.

- Add:**  $2.32 + 71.4 + 0.003 =$   
 (a) 73.75  
 (b) 94.9  
 (c) 9.49  
 (d) 73.723
- Add:**  $7.4 + 0.7 + 1.7 + 1.9 =$   
 (a) 11.5  
 (b) 11.7  
 (c) 12.1  
 (d) 11.9
- Subtract:**  $7,527 - 149 =$   
 (a) 7,378  
 (b) 7,478  
 (c) 7,388  
 (d) 7,488
- Subtract:**  $11.7 - 12.1 =$   
 (a) +0.4  
 (b) +1.1  
 (c) -0.4  
 (d) +0.4
- Add and Subtract:**  
 $7.2 + 0.9 + 1.8 + 1.6 - 12.2 =$   
 (a) +0.7  
 (b) +23.7  
 (c) -1.7  
 (d) -0.7
- Multiply:**  $300 \times 7.48 =$   
 (a) 2,144  
 (b) 2,244  
 (c) 40.106  
 (d) 292.52
- Multiply:**  $25 \times 75 =$   
 (a) 1875  
 (b) 2875  
 (c) 18,750  
 (d) 187.5
- Divide:**  $200,000 \div 10,000 =$   
 (a) 2000  
 (b) 200  
 (c) 20  
 (d) 40

**9. Divide:**  $75,000 \div 10,000 =$

- (a) 7.5
- (b) 7.0
- (c) 75
- (d) 10

**10.** You have a pool that is 60 feet in length and 30 feet in width. How many square feet of surface area does this pool have:

- (a) 2,800 square feet
- (b) 3,600 square feet
- (c) 6,000 square feet
- (d) 1,800 square feet

**11.** The volume of your pool is 328,637 gallons. What is the volume rounded to the nearest thousand?

- (a) 328
- (b) 329
- (c) 329,000
- (d) 328,000

**12.** The current chlorine reading in your pool is 1.5 ppm. You want to raise it to 3.0 ppm. How many more ppm of chlorine do you need to add?

- (a) 4.5 ppm
- (b) 3.0 ppm
- (c) 1.5 ppm
- (d) 2.0 ppm

**13.** How many cubic yards of concrete are needed to make a cement floor of a spa that 9 feet x 12 feet and 6 inches thick?

- (a) 2
- (b) 4
- (c) 18
- (d) 54

**14.** Your pool slopes from 3.5 feet to 6.5 feet. What is the average depth of this pool?

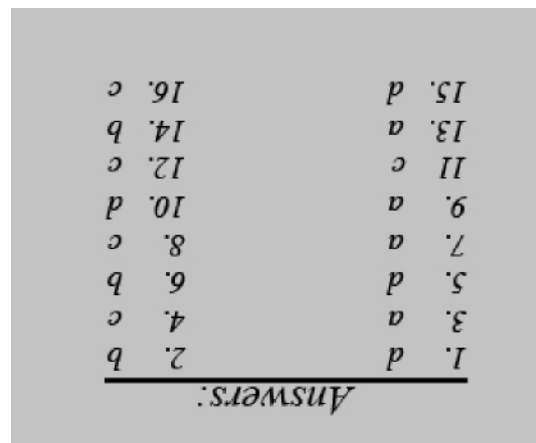
- (a) 10 feet
- (b) 5 feet
- (c) 22.75 feet
- (d) 3 feet

**15.** There is a leak in your pool and it loses 2.5 inches of water each day. It takes 1235 gallons for each inch of water in your 60 feet x 30 feet pool. How many gallons do you need to add to your pool each day?

- (a) 494 gallons
- (b) 37,050 gallons
- (c) 4,500 gallons
- (d) 3,087.5 gallons

**16.** The diameter of a spa is 24 feet. What is the radius?

- (a) 6 feet
- (b) 18 feet
- (c) 12 feet
- (d) 24 feet



## **CALCULATION FORMULAS**

<b>AMOUNT CONVERSIONS</b>	
Ounces to Pounds	Ounces ÷ 16 = <b>Pounds</b>
Fluid Ounces to Gallons	Fluid Ounces ÷ 128 = <b>Gallons</b>

<b>DISTANCE CONVERSIONS</b>	
Yards to Feet	Yards <b>X 3 = Feet</b>
Meters to Feet	Meters <b>X 3.28 = Feet</b>

<b>SURFACE AREA</b>	
Rectangle or Square	Length <b>X Width = Square Feet</b>
Circle ( $\pi r^2$ )	Radius <b>X Radius X 3.14 = Square Feet</b> (Radius = Diameter ÷ 2)

<b>POOL WATER VOLUME</b>	
Average Depth	(Shallow + Deep) ÷ 2
Water Volume	Surface Area <b>X Avg. Depth X 7.5</b>

<b>TURNOVER &amp; FLOW RATE</b>	
Turnover Rate (TOR in Hours)	Water Volume ÷ Flow Rate ÷ 60
Flow Rate (FR in GPM)	Water Volume ÷ TOR ÷ 60

<b>FILTER SIZING</b>	
Filter Area (FA in Sq. Ft.)	Flow Rate ÷ Filter Media Rate
Filter Media Rate (FMR)	Flow Rate ÷ Filter Area
Flow Rate (FR in GPM)	Filter Area <b>X FMR</b>

<b>HEATER SIZING</b>	
BTU	Water Volume <b>X 8.33 X Temp. Rise</b>
Time to reach Temp. (in Hours) (Assuming no heat loss in plumbing or pool)	BTUs ÷ (Heater Rating <b>X Efficiency Rating</b> )
Cost (Natural Gas: 1 Therm = 100,000 BTUs)	Time <b>X Heater Rating ÷ 100,000 X Therm Rate</b>

# Langelier SATURATION INDEX

$$\text{SI} = \text{pH} + \text{Tf} + \text{Cf} + \text{Af} - \text{TDSf}$$

pH as Measured      Temp Factor      Calcium Factor      Total Carbonate Alkalinity Factor      Total Dissolved Solids Factor

## Saturation Index Factors

Temperature		Calcium Hardness <small>Expressed as CaCO3</small>		Total Carbonate Alkalinity		Total Dissolved Solids	
Temp °F	Tf	ppm	Cf	ppm	Af	ppm	TDSf
32	0.0	25	1.0	25	1.4	< 800	12.1
37	0.1	50	1.3	50	1.7	801-1,500	12.2
46	0.2	75	1.5	75	1.9	1,501-2900	12.3
53	0.3	100	1.6	100	2.0	2901-5,500	12.4
60	0.4	125	1.7	125	2.1	> 5,500	12.5
66	0.5	150	1.8	150	2.2		
76	0.6	200	1.9	200	2.3		
84	0.7	250	2.0	250	2.4		
94	0.8	300	2.1	300	2.5		
105	0.9	400	2.2	400	2.6		
		800	2.5	800	2.9		

If an actual measurement is not found in the chart, use the next greatest value.

Temperature		Calcium Hardness		Carbonate Alkalinity		Total Dissolved Solids	
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		800	2.5	800	2.9		

# SATURATION INDEX

## WORKSHEET

	Value	Factor	New Value	Factor
pH				
Temperature				
Calcium Hardness				
Carbonate Alkalinity				
<b>Sub-Total</b>				
Total Dissolved Solids				
<b>Saturation Index</b>				

	Value	Factor	New Value	Factor
pH				
Temperature				
Calcium Hardness				
Carbonate Alkalinity				
<b>Sub-Total</b>				
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## WORKSHEET

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<b>Temperature</b>				
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<b>Carbonate Alkalinity</b>				
<b>Sub-Total</b>				
<b>Total Dissolved Solids</b>				
<b>Saturation Index</b>				

	Value	Factor	New Value	Factor
<b>pH</b>				
<b>Temperature</b>				
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# SATURATION INDEX WORKSHEET

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pH				
Temperature				
Calcium Hardness				
Carbonate Alkalinity				
<b>Sub-Total</b>				
Total Dissolved Solids				
<b>Saturation Index</b>				

# Water Chemistry Guidelines

## The Balance

Parameter	Min.	Ideal	Max.	Pool Type
pH	7.2	7.4 - 7.6	7.8	All Types
Total Alkalinity (ppm)	60	80 – 100*	180	All Types
		100 –120**		
Calcium Hardness (ppm)	150	200 - 400	1,000	Pools
	100	150 - 250	800	Spas
Total Dissolved Solids (ppm)	NA	NA	1,500 over start-up***	All Types
Cyanuric Acid (ppm)	0	30 – 50	100	Outdoor Pools
	0	0	0	Indoor Pool/Spa
Temperature	78° F	80.5° F	82° F	Competition Pools
	-	-	104° F	Spas
	-	Personal Preference	104° F	Other Pools

\* For calcium hypochlorite, lithium hypochlorite, or sodium hypochlorite

\*\* For sodium dichlor, trichlor, chlorine gas, BCDMH

\*\*\* Start-up includes the TDS contribution of salt found in chlorine generating systems

## The Disinfectant

Parameter	Min.	Ideal	Max.	Pool Type
Free Chlorine (ppm)	1.0	2.0 - 4.0	5.0	Pools
	2.0	3.0 - 5.0	10.0	Spas
Total Bromine (ppm)	2.0	4.0 - 6.0	10.0	All Types

## The Contaminants

Contaminant	Min.	Ideal	Max.	Pool Type
Combined Chlorine (ppm)	0	0	0.2	Pools
	0	0	0.5	Spas
Heavy Metals	None	None	None	All Types
Visible Algae	None	None	None	All Types
Bacteria	None	None	Local Code	All Types

## Chemical Dosages for 10,000 Gallons

Dosages to Treat	10,000 Gallons		
Chemical	Desired Change		
Increase Chlorine	1 ppm	5 ppm	10 ppm
Chlorine Gas	1.3 oz	6.7 oz	13 oz
Calcium Hypochlorite (67%)*	2 oz	10 oz	1.3 lb
Sodium Hypochlorite (12%)	10.7 fl.oz.	1.7 qtrs.	3.3 qtrs.¹
Lithium Hypochlorite	3.8 oz	1.2 lbs	2.4 lbs
DiChlor (62%)	2.1 oz	10.75 oz	1.3 lbs
DiChlor (56%)	2.4 oz	12 oz	1.4 lbs
TriChlor	1.5 oz	7.5 oz	14 oz

Increase Total Alkalinity	10 ppm	30 ppm	50 ppm
Sodium Bicarbonate	1.4 lbs	4.2 lbs	7.0 lbs
Sodium Carbonate	14 oz	2.6 lbs	4.4 lbs
Sodium Sesquicarbonate	1.25 lbs	3.75 lbs	6.25 lbs

Decrease Total Alkalinity	10 ppm	30 ppm	50 ppm
Muriatic Acid (31.4%)	26 fl.oz.	2.4 qtrs.	1 gal
Sodium Bisulfate	2.1 lbs	6.4 lbs	10.5 lbs

Increase Calcium Hardness	10 ppm	30 ppm	50 ppm
Calcium Chloride (100%)	.9 lbs	2.8 lbs	4.6 lbs
Calcium Chloride (77%)	1.2 lbs	3.6 lbs	6.0 lbs

Increase Stabilizer	10 ppm	30 ppm	50 ppm
Cyanuric Acid	13 oz	2.5 lbs	4.1 lbs

Neutralize Chlorine	1 ppm	5 ppm	10 ppm
Sodium Thiosulfate	2.6 oz	13 oz	1.6 lbs
Sodium Sulfite	2.4 oz	12 oz	1.5 lbs

Chemical amounts have been rounded off for convenience. Always follow the instructions on the manufacturer's label for exact amounts.

\* **Other** calcium hypochlorite products are available from 47% to 78%.  
 Remember to follow the label directions for dosage amounts.

*(Pool & Hot Tub Alliance, 2023 Pool & Spa Handbook, Appendix B-2)*

## Chemical Adjustment Worksheet

A – Amount (from dosage chart or other source)	B – BIG		C – Change		Total
	<u>B-1</u> Actual Gals		<u>C-1</u> Actual Chg		
		Divide		Divide	
	<u>B-2</u> Label Gals		<u>C-2</u> Label Chg		
	Times		Times		
				Divide ? <b>oz?</b> (16) or <b>fl. oz?</b> (128)	
				Final Answer	

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# Breakpoint Super-Chlorination Worksheet

**Step One: Determine the amount of Combined Chlorine** **(CC = TC – FC)**

**Step Two: Calculate the Breakpoint Chlorination (BPC) amount** **(CC x 10)**

**Step Three: Determine the desired change amount (DC)** **(BPC – FC)**

	<b>Total Chlorine</b>		<b>Free Chlorine</b>		<b>Combine Chlorine</b>
<b>Step One</b>		-		=	
<b>Step Two</b>	<b>Multiply by 10</b>				
<b>Step Three</b>	<b>Subtract the Free Chlorine</b>				-
<b>Use this number in the C-1 (desired change) blank in the Chemical Adjustment Worksheet</b>					

<b>Chemical Adjustment Worksheet</b>					
<b>A – Amount</b> (from dosage chart or other source)	<b>B – BIG</b>		<b>C – Change</b>		<b>Total</b>
	<u><b>B-1</b></u> Actual Gals		<u><b>C-1</b></u> Actual Chg		
		Divide		Divide	
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				Final Answer	

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**Step Two: Calculate the Breakpoint Chlorination (BPC) amount** (CC x 10)

**Step Three: Determine the desired change amount (DC)** (BPC – FC)

	Total Chlorine	-	Free Chlorine	=	Combine Chlorine
<b>Step One</b>		-		=	
<b>Step Two</b>	<b>Multiply by 10</b>				
<b>Step Three</b>	<b>Subtract the Free Chlorine</b>				<b>-</b>
<b>Use this number in the C-1 (desired change) blank in the Chemical Adjustment Worksheet</b>					

<b>Chemical Adjustment Worksheet</b>					
<b>A – Amount</b> (from dosage chart or other source)	<b>B – BIG</b>		<b>C – Change</b>		
	<u>B-1</u> Actual Gals		<u>C-1</u> Actual Chg		
		Divide		Divide	
	<u>B-2</u> Label Gals		<u>C-2</u> Label Chg		
	Times		Times		
				Divide ? <b>oz?</b> (16) or <b>fl. oz?</b> (128)	
				Final Answer	

# Breakpoint Super-Chlorination Worksheet

**Step One: Determine the amount of Combined Chlorine** (CC = TC – FC)

**Step Two: Calculate the Breakpoint Chlorination (BPC) amount** (CC x 10)

**Step Three: Determine the desired change amount (DC)** (BPC – FC)

	Total Chlorine	-	Free Chlorine	=	Combine Chlorine
<b>Step One</b>		-		=	
<b>Step Two</b>	<b>Multiply by 10</b>				
<b>Step Three</b>	<b>Subtract the Free Chlorine</b>				<b>-</b>
<b>Use this number in the C-1 (desired change) blank in the Chemical Adjustment Worksheet</b>					

<b>Chemical Adjustment Worksheet</b>					
<b>A – Amount</b> (from dosage chart or other source)	<b>B – BIG</b>		<b>C – Change</b>		
	<u>B-1</u> Actual Gals		<u>C-1</u> Actual Chg		
		Divide		Divide	
	<u>B-2</u> Label Gals		<u>C-2</u> Label Chg		
	Times		Times		
				Divide ? <b>oz?</b> (16) or <b>fl. oz?</b> (128)	
				Final Answer	

# Breakpoint Super-Chlorination Worksheet

**Step One: Determine the amount of Combined Chlorine** (CC = TC – FC)

**Step Two: Calculate the Breakpoint Chlorination (BPC) amount** (CC x 10)

**Step Three: Determine the desired change amount (DC)** (BPC – FC)

	Total Chlorine	-	Free Chlorine	=	Combine Chlorine
Step One		-		=	
Step Two	Multiply by 10				
Step Three	Subtract the Free Chlorine				-
Use this number in the C-1 (desired change) blank in the Chemical Adjustment Worksheet					

Chemical Adjustment Worksheet					
A – Amount (from dosage chart or other source)	B – BIG		C – Change		
	<u>B-1</u> Actual Gals		<u>C-1</u> Actual Chg		
		Divide		Divide	
	<u>B-2</u> Label Gals		<u>C-2</u> Label Chg		
	Times		Times		
				Divide ? <b>oz?</b> (16) or <b>fl. oz?</b> (128)	
				Final Answer	

The following pages contain  
additional information  
which is not in  
or explicitly stated in the

*Pool & Spa Operator Handbook*™

# PERIODIC TABLE OF THE ELEMENTS

<http://www.periodni.com>

GROUP	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
PERIOD	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1	1 <b>H</b> HYDROGEN	2 <b>He</b> HELIUM																	
2	3 <b>Li</b> LITHIUM	4 <b>Be</b> BERYLLIUM	5 <b>B</b> BORON	6 <b>C</b> CARBON	7 <b>N</b> NITROGEN	8 <b>O</b> OXYGEN	9 <b>F</b> FLUORINE	10 <b>Ne</b> NEON											
3	11 <b>Na</b> SODIUM	12 <b>Mg</b> MAGNESIUM	13 <b>Al</b> ALUMINIUM	14 <b>Si</b> SILICON	15 <b>P</b> PHOSPHORUS	16 <b>S</b> SULPHUR	17 <b>Cl</b> CHLORINE	18 <b>Ar</b> ARGON											
4	19 <b>K</b> POTASSIUM	20 <b>Ca</b> CALCIUM	21 <b>Sc</b> SCANDIUM	22 <b>Ti</b> TITANIUM	23 <b>V</b> VANADIUM	24 <b>Cr</b> CHROMIUM	25 <b>Mn</b> MANGANESE	26 <b>Fe</b> IRON	27 <b>Co</b> COBALT	28 <b>Ni</b> NICKEL	29 <b>Cu</b> COPPER	30 <b>Zn</b> ZINC	31 <b>Ga</b> GALLIUM	32 <b>Ge</b> GERMANIUM	33 <b>As</b> ARSENIC	34 <b>Se</b> SELENIUM	35 <b>Br</b> BROMINE	36 <b>Kr</b> KRYPTON	
5	37 <b>Rb</b> RUBIDIUM	38 <b>Sr</b> STRONTIUM	39 <b>Y</b> YTTRIUM	40 <b>Zr</b> ZIRCONIUM	41 <b>Nb</b> NIObIUM	42 <b>Mo</b> MOlyBDENUM	43 <b>Tc</b> TECHNETIUM	44 <b>Ru</b> RUTHENIUM	45 <b>Rh</b> RHODIUM	46 <b>Pd</b> PALLADIUM	47 <b>Ag</b> SILVER	48 <b>Cd</b> CADMIUM	49 <b>In</b> INDIUM	50 <b>Sn</b> TIN	51 <b>Sb</b> ANTIMONY	52 <b>Te</b> TELLURIUM	53 <b>I</b> IODINE	54 <b>Xe</b> XENON	
6	55 <b>Cs</b> CAESIUM	56 <b>Ba</b> BARIUM	57-71 <b>La-Lu</b> Lanthanide	72 <b>Hf</b> HAFNIUM	73 <b>Ta</b> TANTALUM	74 <b>W</b> TUNGSTEN	75 <b>Re</b> RHENIUM	76 <b>Os</b> OSMIUM	77 <b>Ir</b> IRIDIUM	78 <b>Pt</b> PLATINUM	79 <b>Au</b> GOLD	80 <b>Hg</b> MERCURY	81 <b>Tl</b> THALLIUM	82 <b>Pb</b> LEAD	83 <b>Bi</b> BISMUTH	84 <b>Po</b> POLONIUM	85 <b>At</b> ASTATINE	86 <b>Rn</b> RADON	
7	87 <b>Fr</b> FRANCIUM	88 <b>Ra</b> RADIUM	89-103 <b>Ac-Lr</b> Actinide	104 <b>Rf</b> RUTHERFORDIUM	105 <b>Db</b> DUBNIUM	106 <b>Sg</b> SEABORGIUM	107 <b>Bh</b> BOHRRIUM	108 <b>Hs</b> HASSIUM	109 <b>Mt</b> MEITNERIUM	110 <b>Ds</b> DARMSTADTIUM	111 <b>Rg</b> ROENTGENIUM	112 <b>Cn</b> COPERNICIUM	113 <b>Uut</b> UNUNTRIUM	114 <b>Ff</b> FLEROVIUM	115 <b>Uup</b> UNUNPENTIUM	116 <b>Lv</b> LIVERMORIUM	117 <b>Uus</b> UNUNSEPTIUM	118 <b>Uuo</b> UNUNOCTIUM	

Legend for element categories:

- Metal (Blue)
- Semimetal (Light Blue)
- Nonmetal (Green)
- Alkali metal (Light Blue)
- Alkaline earth metal (Light Blue)
- Chalcogens element (Light Green)
- Halogens element (Light Green)
- Noble gas (Light Green)
- Transition metals (Light Blue)
- Lanthanide (Light Purple)
- Actinide (Light Purple)

STANDARD STATE (25 °C; 101 kPa):

- Ne - gas
- Hg - liquid
- Fe - solid
- Tc - synthetic

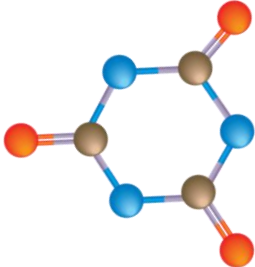
Copyright © 2012 Eni Generalic

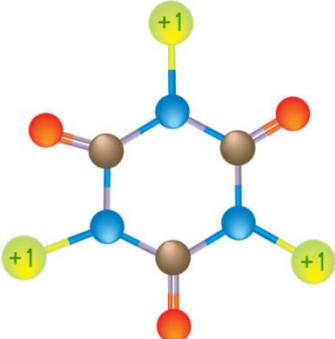
GROUP	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
PERIOD	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75
<b>La</b> LANTHANUM	<b>Ce</b> CERIUM	<b>Pr</b> PRASEODYMIUM	<b>Nd</b> NEODYMIUM	<b>Pm</b> PROMETHIUM	<b>Sm</b> SAMARIUM	<b>Eu</b> EUROPIUM	<b>Gd</b> GADOLINIUM	<b>Tb</b> TERBIUM	<b>Dy</b> DYSPROSIUM	<b>Ho</b> HOLMIUM	<b>Er</b> ERBIUM	<b>Tm</b> THULIUM	<b>Yb</b> YTTERIUM	<b>Lu</b> LUTETIUM				
89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107
<b>Ac</b> ACTINIUM	<b>Th</b> THORIUM	<b>Pa</b> PROTACTINIUM	<b>U</b> URANIUM	<b>Np</b> NEPTUNIUM	<b>Pu</b> PLUTONIUM	<b>Am</b> AMERICIUM	<b>Cm</b> CURIUM	<b>Bk</b> BERKELIUM	<b>Cf</b> CALIFORNIUM	<b>Es</b> EINSTEINIUM	<b>Fm</b> FERMIUM	<b>Md</b> MENDELEVIUM	<b>No</b> NOBELIUM	<b>Lr</b> LAWRENCIUM				

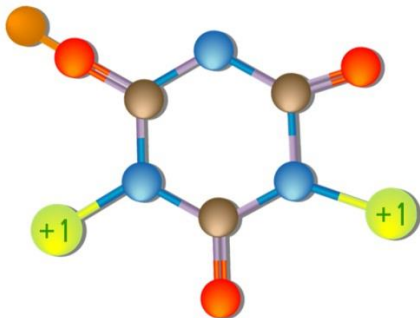
LANTHANIDE

(1) Pure Appl. Chem., 81, No. 11, 2131-2156 (2009)  
 Relative atomic masses are expressed with five significant figures. For elements that have no stable nuclides, the value enclosed in brackets indicates the mass number of the longest-lived isotope of the element. However three such elements (Th, Pa and U) do have a characteristic terrestrial isotopic composition, and for these an atomic weight is tabulated.

<u>Element</u>	<u>Atomic Weight</u>	<h1>Atomic Weights Of Organic Chlorine</h1>
Carbon	12.01	
Nitrogen	14.01	
Oxygen	16	
Sodium	22.99	
Chlorine	35.45	

<u>Cyanuric Acid</u>		$C_3N_3O_3$
Carbon	36.03	
Nitrogen	42.03	
Oxygen	48	
<u>Total</u>	<u>126.06</u>	

<u>TriChlor</u>		$C_3N_3O_3Cl_3$
CYA	126.06	
Chlorine (3)	106.35	
<u>Total</u>	<u>232.41</u>	
%Chlor	<b>45.76%</b>	
%CYA	54.24%	

<u>DiChlor</u>		$NaCl_2C_3N_3O_3$
CYA	126.06	
Sodium (1)	22.99	
Sub-Total	149.05	
Chlorine (2)	70.9	
<u>Total</u>	<u>219.95</u>	
%Chlor	<b>32.23%</b>	
%CYA	57.31%	
%Sodium	10.45%	

# Cyanuric Acid

(Chlorine Stabilizer / Pool Water Conditioner)

Cyanuric acid (CYA) is used in pool water to protect the chlorine disinfectant from the sun's UV. It does this by forming a weak molecular bond with the disinfectant, thus keeping it in the water longer. This bond, however, **does slow down** the killing rate of the chlorine disinfectant.

To ensure there is enough active chlorine disinfectant to keep a healthy swimming pool and to prevent algae when using CYA, the measured FC level should be at least **7.5 percent** of the amount of CYA. For example: if the CYA level is 50ppm then the measured residual FC should be at least 3.75 (4ppm).

High levels of CYA -- i.e., 100, 200 and higher -- will require impractical high levels of chlorine disinfectant to prevent algae and to ensure the inactivation of harmful bacteria and other pathogens that may be introduced into the swimming pool. When CYA levels are this high, then supplemental chemicals (algaecides, oxidizers, phosphate removers, etc.) can be used to help prevent algae and to help keep the pool safe and healthy.

Table 5 is the guideline that shows the amount of chlorine disinfectant needed as the CYA levels go up, given an equivalent killing rate. These chlorine/CYA ratios should be followed to maintain a healthy pool.

<b>Chlorine / CYA Chart</b>				
<b>CYA (ppm)</b>	<b>Free available Chlorine (ppm)</b>			
	<b>Minimum</b> (≈7.5% of CYA)	<b>Target</b> (≈11.5% of CYA)	<b>Shock/Algae</b> (≈40% of CYA)	<b>Yellow Algae Kill</b> (≈60% of CYA)
0	.07 <sup>1</sup>	.1 <sup>1</sup>	.7 <sup>1</sup>	2 <sup>1</sup>
10	1 <sup>1</sup>	1.5 <sup>1</sup>	5	7
20	2	3	10	13
30	2	4	12	18
40	3	5	16	24
50	4	6	20	30 <sup>2</sup>
60	5	7	24	35 <sup>2</sup>
70	5	8	28 <sup>2</sup>	41 <sup>2</sup>
80	6	9	31 <sup>2</sup>	46 <sup>2</sup>
90	7	10	35 <sup>2</sup>	52 <sup>2</sup>
100	7	12	39 <sup>2</sup>	58 <sup>2</sup>
120	9	14	47 <sup>2</sup>	68 <sup>2</sup>

Table 5

<sup>1</sup>A minimum FC level is needed as a "reserve" for usage so in practice at least 2 ppm FC is required even at low CYA levels. The table above shows the amount needed for disinfecting chlorine for equivalent killing power (rates), but does not take into account the amount needed in reserve to prevent getting used up as this varies by pool.

<sup>2</sup>The shock levels shown have equivalent disinfecting chlorine amounts (in a column) but at high CYA levels it may be impractical to use such high FC levels. A partial drain/refill to lower the CYA level is usually what is needed, or one can shock at a lower level but will take longer to kill the algae.

**ACKNOWLEDGMENT:** The chlorine/CYA chart was developed by Ben Powell of 'The Pool Forum' and Richard Falk of 'Trouble Free Pool'.



## Universal Dosage Formula

$$\text{Pounds} = \left( \frac{\text{Pool Gallons} \times \text{PPM change}}{120,000} \right) \times \text{Multiplier}$$

Chemical	Multiplier		Weight in 1lb DE Scoop
<b>Alkalinity – Down</b>			
Muriatic Acid ( <b>quarts</b> )	.96		N/A
Sodium Bisulfate (Dry Acid)	2.55		4.6 lbs.
<b>Alkalinity – Up</b>			
Sodium Bicarb (Baking Soda)	1.68		3.9 lbs.
Sodium Carbonate (Soda Ash)	1.06		4 lbs.
<b>Chlorine</b>		<b>Amount Needed to Equal ACC of 1 lb. of Cl<sub>2</sub> gas</b>	
Sodium Hypochlorite - Bleach (Liquid 12%, <b>gallon</b> )	1	<b>1 gallon</b>	N/A
TriChlor (90%)	1.09	1.1 lbs. (≈ 2 tabs)	3.5 lbs.
Sodium DiChlor (56%)	1.81	1.8 lbs.	3.3 lbs.
Calcium Hypochlorite (65%)	1.55	1.5 lbs.	2.9 lbs.
<b>Calcium - Calcium Chloride (77%)</b>	1.44		3 lbs.
<b>Stabilizer - Cyanuric Acid</b>	1		2.4 lbs.
<b>Salt</b>	1		3.8 lbs.
<b>Borate</b> (5 mol, pentahydrate)	6.74		
(10 mol, decahydrate)	9.09		
Boric Acid	5.7		

$$\text{PPM} = \frac{\text{Pounds} \times 120,000}{\text{Pool Gallons/Multiplier}}$$

$$\text{Pool Gallons} = \frac{\text{Pounds} \times 120,000}{\text{PPM/Multiplier}}$$

## Equivalents of Common Pool Chemicals

Chemical Name	Equivalent to 1 lb. of Cl <sub>2</sub> Gas	Weight in a 1 lb. DE Scoop	Price per Pound 06-2013 / 01-2023
Sodium Hypochlorite (Bleach)	1 Gallon 12% Trade Grade	N/A	\$2.86/Gallon <b>\$5.66</b>
Sodium DiChlor (56%)	1.8 lbs.	3.3 lbs.	\$2.15 / <b>\$5.12</b>
TriChlor 3" Tabs (90%)	1.1 lbs. ( ≈ 2 tabs)	N/A	\$1.88 / <b>\$4.10</b>
TriChlor (Granular) (90%)	1.1 lbs.	3.5 lbs.	\$2.38 / <b>\$6.00</b>
Calcium Hypochlorite (65%)	1.5 lbs.	2.9 lbs.	\$1.63 / <b>\$2.72</b>
D.E.		1 lb.	\$0.33 / <b>\$0.54</b>
Cyanuric Acid (Chlorine Stabilizer)		2.4 lbs.	\$0.85 / <b>\$1.44</b>
Salt		3.8 lbs.	\$0.15 / <b>\$0.28</b>
Sodium Bicarbonate (Baking Soda)		3.9 lbs.	\$0.45 / <b>\$0.64</b>
Sodium Carbonate (Soda Ash)		4 lbs.	\$0.44 / <b>\$0.44</b>
Potassium Monopersulfate		4.3 lbs.	\$3.34 / <b>\$5.04</b>
Sodium Bromide		5.2 lbs.	\$3.96 / <b>\$5.44</b>
Muriatic Acid (Liquid Acid)		N/A	\$4.18/Gallon <b>\$7.50</b>
Sodium Bisulfate (Dry Acid)	<b>Equivalent to 1 quart. Muriatic Acid (TA)</b>	4.6 lbs.	\$1.40 / <b>\$1.53</b>
	≈ 2.5 pounds		
Boric Acid			<b>\$0.96</b>

## pH of Common Pool Water Chemicals

<b>Chemical</b>	<b>pH</b>
Sodium Carbonate (Soda Ash)	≈ 12.3 or higher
Sodium Hypochlorite (Liquid Chlorine, Bleach)	9-13
Calcium Hypochlorite (Cal Hypo)	8.5-11
Sodium Tetraborate Pentahydrate (Endure)	9.1-9.2
Calcium Chloride	8-9
Sodium Bicarbonate (Baking Soda, BiCarb)	8.3
Sodium Bromide	6.5-8
DiChlor	6.8-7.1
Boric Acid	5.1
Cyanuric Acid (Chlorine Stabilizer, Conditioner)	3-4
TriChlor	2.7-2.9
Potassium Monopersulfate	2-2.3
Sodium Bisulfate (Dry Acid)	1.4
Muriatic Acid	< 1

## **Adding Chlorine Compounds to The Swimming Pool**

When chlorine compounds are added to a swimming pool to disinfect and oxidize the water, the active chlorine (HOCl) gets used up and the other elements of the compound will build up. When the active chlorine does its job and gets used/consumed, it converts to chloride, i.e. salt which also builds up in the pool water.

The following chart shows the rate of build-up of salt and other chemicals per 10ppm of chlorine added to a swimming pool using the various chlorine compounds.

<b>Chlorine Compound (10 ppm)</b>	<b><u>Salt</u> Build-Up (in ppm)</b>	<b><u>Chemical</u> Build-Up (in ppm)</b>
Sodium Hypochlorite (10 ppm)	<b>16.4 ppm</b>	
TriChlor (10ppm)	<b>8.2 ppm</b>	Cyanuric Acid <b>6.1 ppm</b>
DiChlor (10ppm)	<b>8.2 ppm</b>	Cyanuric Acid <b>9.1 ppm</b>
Calcium Hypochlorite (10ppm)	<b>8.2 ppm</b>	Calcium Hardness <b>7 ppm</b>

**ACKNOWLEDGMENT:** These ratios are published online at the forum: [TroubleFreePool.com](http://TroubleFreePool.com), by Richard Falk, aka “ChemGeek”. He has also published them on various other forums and articles that he has written.

## Incompatible Chemicals

<b>Incompatible Chemicals</b>	<b>Result</b>
Calcium Chloride (Hardness Increaser) <b>AND</b> Sodium Bicarbonate (BiCarb) OR Sodium Carbonate (Soda Ash)	Calcium Chloride and bicarb or soda ash should not be added at the same time or even within a few hours of each other. A white precipitate will form, clouding the water and may cause scaling
TriChlor <b>AND</b> Cal Hypo	Explosion and Fire (by themselves)
Chlorine – Any kind <b>AND</b> Iron	Stain and discolored water. Iron above .5ppm will rapidly destroy the chlorine.
Polymer water Clarifier <b>AND</b> Sequestering agent	Cloudiness that may not go away or filter out
Phosphate Remover <b>AND</b> Sequestering Agent	Cloudiness
Phosphate Remover <b>AND</b> High level of calcium or magnesium	Cloudiness and a precipitate that may scale

## RECREATIONAL WATER ILLNESSES AT A GLANCE

<b>PATHOGAN</b>	<b>TYPE</b>	<b>METHOD OF TRANSMISSION</b>	<b>SYMPTOMS</b>	<b>REAL WORLD OCCURRENCE</b>	<b>CHLORINATION TIME (1PPM FAC)</b>
Cryptosporidium	Parasite	Swallowing contaminated water. Very contagious, through people-to-people contact.	Dehydration, weight loss, stomach cramps, fever, nausea, vomiting. No treatment.	In 2008, several pools and water parks in the Dallas/Ft. Worth area were contaminated and closed. Hundreds of people were sick, one died	About 255 hours (10.6 days)
E.coli	Bacteria	Swallowing contaminated water.	Severe bloody diarrhea, abdominal cramps, kidney failure. Antibiotics available.	In 1998, 26 children fell ill from an outbreak in a Marietta, Georgia wading pool. Seven had kidney failure; one died.	Less than one minute
Giardiasis	One-celled parasite	Swallowing contaminated water. Cysts can survive in cold water for months.	Diarrhea, gas, greasy stools, stomach cramps, upset stomach, nausea. Prescription drugs available.	In 2003, 55 people were struck at a country club in Milton, Massachusetts. The source believed to be the children's swimming pool.	About 45 minutes
Hepatitis	A Virus	Swallowing water contaminated with feces infected with virus.	Jaundice, fatigue, stomach pain, loss of appetite, nausea, diarrhea, fever. No real treatment. Vaccine available.	The public pools at a campground in Louisiana were the source of a multistate outbreak in 1989. 20 people were infected.	About 16 minutes
Legionnaires' Disease	Bacterium Legionella pneumophila	Breathing mists from hot tubs infected with the bacteria. Not contagious.	Fever, chills, cough, muscle aches, headache, fatigue, diarrhea, kidney malfunction. Legionnaire's treatable.	Over 120 people became ill after attending a conference at the Playboy Mansion in L.A. in February 2011. Legionella pneumophila bacteria was found in the grotto spa	Less than one minute
Naegleria Infection	Microscopic amoeba	Enters through the nose and travels to brain and spinal cord. Feeds on brain tissue. Very rare infections.	Causes primary amebic meningoencephalitis (PAM), a brain inflammation. Drugs available, high fatality rate.	In 1978, a 9-year-old girl in San Francisco was infected in a hot springs pool; she was one of only three known survivors of this disease.	Less than one minute
Norovirus Gastroenteritis	Virus	Swallowing water contaminated with feces infected with virus. Very contagious.	Nausea, vomiting, diarrhea, stomach cramping, fever, chills, muscle aches, fatigue. Most recover in 1-2 days.	In 2004, 53 people fell ill at a swim club in Vermont.	About 30 minutes to an hour
Pseudomonas Dermatitis (Hot tub rash)	Bacteria, Pseudomonas Aeruginosa	Direct skin contact with contaminated water. Usually in hot tubs but also in pools.	Itchy skin, red rash, blisters around hair follicles. Clears up on its own in a few days.	Nine cases were documented at a hotel pool and spa in Bangor, Maine. In January 2009, this bacterium was found in the blood of a Brazilian Model – both hands and feet were amputated – she died	Less than one minute
Salmonellosis	Bacteria, Salmonella	Swallowing water contaminated with bacteria.	Diarrhea, fever, cramps. Antibiotics if infection spreads to intestines; no necessary treatment otherwise.	Three cases were documented at a park pool in 1995.	Less than one minute
Shigellosis Shigella	Bacteria, Shigella	Swallowing water contaminated with bacteria.	Bloody diarrhea, fever, cramps. Antibiotics-though forms of shigella have become resistant.	An un-chlorinated wading pool in Dubuque, Iowa, caused at least 69 cases in 2001	Less than one minute

Courtesy Aquatics International Magazine